Tetrahedron 65 (2009) 3003–3013

Contents lists available at [ScienceDirect](www.sciencedirect.com/science/journal/00404020)

Tetrahedron

journal homepage: www.elsevier.com/locate/tet

Anti-Pseudomonas aeruginosa xanthones from the resin and green fruits of Cratoxylum cochinchinense

Nawong Boonnak ^{a, b}, Chatchanok Karalai ^{a, b,} *, Suchada Chantrapromma ^a, Chanita Ponglimanont ^b, Hoong-Kun Fun ^c, Akkharawit Kanjana-Opas ^d, Kan Chantrapromma ^e, Shigeru Kato ^f

a Crystal Materials Research Unit, Department of Chemistry, Faculty of Science, Prince of Songkla University, Hat-Yai, Songkhla 90112, Thailand ^b Department of Chemistry and Center of Excellence for Innovation in Chemistry, Faculty of Science, Prince of Songkla University, Hat-Yai, Songkhla 90112, Thailand

^c X-ray Crystallography Unit, School of Physics, Universiti Sains Malaysia, 11800 USM, Penang, Malaysia

^d Department of Industrial Biotechnology, Faculty of Agro-Industry, Prince of Songkla University, Hat-Yai, Songkhla 90112, Thailand

^e Institute of Research and Development, Walailak University, Thasala, Nakhon Si Thammarat, 80160, Thailand

^f Department of Materials and Life Science, Seikei University, Tokyo 180 8633, Japan

article info

Article history: Received 23 October 2008 Received in revised form 8 January 2009 Accepted 22 January 2009 Available online 29 January 2009

Keywords:

Cratoxylum cochinchinense 1,3,7-Trihydroxyxanthone 1,3,7-Trioxygenated xanthone Antifungal activity Antibacterial activity Pseudomonas aeruginosa

1. Introduction

ABSTRACT

Cochinchinones I–L (1–3 and 13) along with 11 known xanthones (4–12, 14, and 15) were isolated from the resin and green fruits of Cratoxylum cochinchinense. In addition, four new acetylated compounds (16– 19) were derivatized from 7-geranyloxy-1,3-dihydroxyxanthone (14) and 3-geranyloxy-1,7-dihydroxyxanthone (15). All compounds were characterized on the basis of spectroscopic analyses. The structures of cochinchinone I (1), a monoacetate (18) and a dibrosylate (20), were also confirmed by X-ray diffraction analysis. The antibacterial and antifungal activities of selected compounds were evaluated as well.

- 2009 Elsevier Ltd. All rights reserved.

Cratoxylum cochinchinense belongs to the family Guttiferae, which is distributed in several parts of Thailand.¹ This plant is a well-known tropical tree, and is commonly known in Thailand as Tui Kliang. The bark, roots, and leaves of this plant have been used in folk medicine to treat fever, coughs, diarrhea, itches, ulcers, and abdominal complaints.² Previous investigations revealed the major components from the Cratoxylum genus as xanthones and anthraquinones.[3–6](#page-10-0) Xanthones have been reported to exhibit biological activities such as antibacterial, $3-5$ antioxidant, $6,7$ cytotoxic, $3-5,8,9$ and antiprotozoal activities.^{[10](#page-10-0)} A preliminary screening of the bioactivity of the crude extracts from the resin and green fruits of C. cochinchinense has shown strong antibacterial activity specifically against Pseudomonas aeruginosa. Although Pseudomonas sepsis is a major cause of death in transplant recipients, while P. aeruginosa is one of the most common Gram-negative pathogenic bacteria causing life-threatening infections. $11-13$ This result led us to

investigate the bioactive compounds from this plant. The antibacterial and antifungal activities of isolated compounds were also evaluated.

2. Results and discussion

The crude CH_2Cl_2 extract of the resin of C. cochinchinense was subjected to chemical investigation leading to the isolation of three new xanthones, cochinchinones I–K (1–3), together with nine known xanthones identified as cochinchinone A (4) , $(6\ 1,3,7$ $(6\ 1,3,7$ $(6\ 1,3,7$ -trihydroxy-2,4-diisoprenylxanthone (5) ,¹⁴ celebixanthone methyl ether (6),^{[15](#page-10-0)} dulxis-xanthone F (7),¹⁵ β -mangostin (8),^{[16](#page-10-0)} α -mangostin (9) ,¹⁶ macluraxanthone (10) ,¹⁷ pruniflorone G (11) ,^{[3](#page-10-0)} and a cagedprenylated xanthone (12) ⁶ Their structures were elucidated by NMR analysis and comparison of their spectroscopic data with those reported in the literature.

Cochinchinone I (1) was obtained as a yellow powder, which was further recrystallized from acetone to yield yellow single crystals. The X-ray structure [\(Fig. 1](#page-1-0)) suggested a xanthone skeleton with a chromene ring. A molecular formula $C_{28}H_{30}O_5$ was implied by HRMS. Its structure was supported by the 1 H and 13 C NMR spectral data ([Table 1](#page-1-0)). The $^1\mathrm{H}$ and $^{13}\mathrm{C}$ NMR spectral data of **1** were

Corresponding author. Tel.: +66 7428 8444; fax: +66 7421 2918.

E-mail addresses: [chatchanok.k@psu.ac.th,](mailto:chatchanok.k@psu.ac.th) chatchanok_k@yahoo.com (C. Karalai).

^{0040-4020/\$ –} see front matter © 2009 Elsevier Ltd. All rights reserved. doi:10.1016/j.tet.2009.01.083

Figure 1. The X-ray structure of 1.

comparable to cochinchinone A (4), which was previously isolated from the roots of C. cochinchinense.^{[6](#page-10-0)} The main difference was observed at C-2 and C-3, where the $^1\mathrm{H}$ NMR spectral data of **1** showed the signals of a chromene ring at δ 6.74 (d, J 9.9 Hz, H-1'), 5.60 (d, J 9.9 Hz, H-2'), and 1.48 (s, Me-4' and Me-5') instead of a prenyl group at C-2 and a free hydroxyl group at C-3 as in 4 (Table 1). The

chromene ring was connected to a xanthone skeleton in a linear fashion whose structure was confirmed by HMBC correlations as summarized in [Table 2,](#page-2-0) which agreed well with the X-ray structure (Fig. 1). From these data, the structure of cochinchinone I was assigned as 1. In addition, the asymmetric unit of a cocrystal (Fig. 1a) consisted of a disordered mixture of cochinchinone I (1) (Fig. 1) and a new chromenoxanthone 1a (Fig. 1b). The latter was different from 1 in which a geranyl side chain in 1 was replaced by a (E) -3",7",7"-trimethyloct-2"-ene side chain in 1a. Therefore, a trace amount of chromenoxanthone 1a in a single crystal should be an unexpected contaminated compound from cochinchinone I (1), which might be transformed during the resin formation.

Cochinchinone J (2) was obtained as a yellow viscous oil with a molecular ion peak at m/z 446.2092 [M]⁺ in the HRMS, corresponding to a molecular formula of $C_{28}H_{30}O_5$. The UV spectrum showed absorption bands of a xanthone (243, 289, 298, 320, 351, and 391 nm), 14 while the IR spectrum exhibited the hydroxyl and conjugated carbonyl functionalities at v_{max} 3397 and 1649 cm⁻¹, respectively. The 1 H and 13 C NMR data of **2** (Table 1) were similar to those of ${\bf 4}^6$ ${\bf 4}^6$ except for the appearance of the signals of a chromene ring bearing a methyl group and six-carbon side chain of 4-methylpent-3 enyl group, which appeared at δ_H 5.10 (br t, J 7.2 Hz, H-6"), 1.89 (m, 1H-4"), 1.68 (m, 1H-4"), 2.12 (m, H₂-5"), 1.66 (s, Me-8"), and 1.57 (s, Me-10") (Table 1) instead of a geranyl moiety at C-4 as in 4. The loss of 4-methylpent-3-enyl moiety in EIMS, m/z 363 ([M]⁺ -83), also supported the proposed structure. Finally, the location of the angular chromene ring was confirmed by HMBC correlations [\(Table 2\)](#page-2-0) in which the methine proton H-1" (δ 6.88) was correlated with C-3 (δ 158.9), C-4 (δ 100.2), C-4a (δ 150.2), and C-3" (δ 80.6). Therefore, the structure of cochinchinone J was deduced to be 2.

Compound 3 was isolated as an inseparable mixture with compound 2 , and the mixture was thus acetylated with $Ac₂O$ in pyridine. The resulting product was further separated by CC eluting with 70% CHCl₃-hexane to give monoacetates $3a(18.9 \text{ mg})$ and $2a$

^a Deduced from HMQC experiment.

(5.1 mg). The latter was confirmed as an acetylated derivative of 2 by comparison of its spectral data with those of compound 2.

Compound 3a is a yellow powder. The HRMS spectrum showed a molecular ion peak at m/z 490.2355 [M]⁺, corresponding to $C_{30}H_{34}O_6$. The UV spectrum showed absorption bands of a xanthone (243, 271, 303, 326, and 383 nm), 14 14 14 while the IR spectrum exhibited the hydroxyl and conjugated carbonyl functionalities at $v_{\rm max}$ 3392 and 1648 cm⁻¹, respectively. The ¹H and ¹³C NMR spectral data of 3a [\(Table 1\)](#page-1-0) were closely related to those of 1, except for the appearance of the signals of a dimethylchromane ring and acetoxy group revealed at $\delta_{\rm H}$ 2.75 (br t, J 6.9 Hz, H-1'), 1.81 (br t, J 6.9 Hz, H-2'), 1.39 (s, Me-4' and Me-5'), and 2.34 (s, 7-OAc) [\(Table 1\)](#page-1-0) instead of a chromene ring and a hydroxyl group in 1. The chromane ring was fused to the xanthone nucleus in a linear fashion, which was confirmed by HMBC correlations (Table 2), in which the methylene protons H₂-1['] at δ_H 2.75 were correlated with C-1 (δ 158.4), C-2 (δ 103.9), C-3 (δ 159.6), C-2' (δ 31.7), and C-3' (δ 76.3), while the H-bonded hydroxyl proton 1-OH (δ 13.02) was correlated with C-1 (δ 158.4), C-2 (δ 103.9), and C-9a (δ 102.3). Therefore, the structure of 3a, an acetylated form of 3, was deduced. Compound 3 was named cochinchinone K.

It is interesting to note that the three new chromenoxanthones (1 and 2) and chromanoxanthone (3) could be derived from cochinchinone A (4). The isoprenyl or geranyl side chains of 4 were epoxidized and further cyclized via a free hydroxyl at C-3 to produce the chromanol rin[g18](#page-10-0) as shown in [Scheme 1.](#page-3-0) Further dehydration led to the linear or angular chromene rings of 1 and 2, respectively. On the other hand, the new chromanoxanthone (3) could be produced by cyclization via a free hydroxyl group to an isoprenyl side chain of 4 to afford the chromane ring ([Scheme 1](#page-3-0)). Therefore, cochinchinone A (4) should be a precursor of cochinchinones I–K (1–3).

The investigation of the crude $CH₂Cl₂$ extract of the green fruits (approximate 4 weeks maturity stage) led to the isolation and identification of a new xanthone, cochinchinone L (13) along with two major known oxygeranyl xanthones ${\bf 14}^{14}$ ${\bf 14}^{14}$ ${\bf 14}^{14}$ and ${\bf 15}^{19}$ ${\bf 15}^{19}$ ${\bf 15}^{19}$ Compounds 14 and 15 were structural isomers of xanthones with an oxygeranyl

Figure 1a. The asymmetric unit of a cocrystal of 1 and 1a. **Figure 15** Figure 1b. The X-ray structure of 1a.

Scheme 1. Plausible biosynthesis pathway of 1, 2, and 3.

side chain. The $^1\mathrm{H}$ and $^{13}\mathrm{C}$ NMR spectral data of both **14** and **15** (Tables 3 and 4) were almost identical. However, the main difference was observed in the NOESY experiment in which H_2-1' (δ 4.45) of 14 showed a cross-peak with H-8 (δ 7.40) and H-6 (δ 7.18), whereas the H₂-1' (δ 4.63) of **15** showed a cross-peak with H-4 (δ

6.40) and H-2 (δ 6.34). These results implied that the oxygeranyl side chain of 14 was connected to C-7 and that of 15 to C-3. To confirm these structural assignments of 14 and 15, compound 14 was further reacted with p-bromobenzenesulfonyl chloride in $CH₂Cl₂$ at room temperature to afford a crystalline dibrosylate 20

Table 3 ¹H NMR (300 MHz) spectral data of **13-19** in CDCl₃

Position	13	14	15	16	17	18	19	
2	6.43, d, 2.1	6.22, d, 1.8	6.34, d, 2.4	6.42, br d, 2.1	7.15, d, 2.1	6.19, br d, 2.1	6.41, d, 2.1	
$\overline{\mathbf{r}}$	6.57, d, 2.1	6.23, d, 1.8	6.40, d, 2.4	6.59, br d, 2.1	6.79, d, 2.1	6.24, br d, 2.1	6.46, d, 2.1	
	7.08, br d, 9.0	7.15, br d, 9.0	7.30, d, 9.3	7.21, br d, 8.1	7.23, d, 9.3	7.25, d, 9.3	7.25, d, 8.1	
$\begin{array}{c} 5 \\ 6 \end{array}$	7.10, br d, 9.0	7.18, br d, 9.0	7.26, d, 9.3	7.20, br d, 8.1	7.28, dd, 9.3, 2.1	7.29, br d, 9.3	7.26, br d, 8.1	
	7.45 , br s	7.40 , br s	7.59 , br s	7.44, br s	7.55, d, 2.1	7.76, br s	7.78, br d, 1.5	
$\begin{array}{c} 8 \\ 1' \end{array}$	4.42, d, 6.3	4.45, d, 6.3	4.63, d, 6.6	4.50, d, 6.6	4.56, d, 6.3	4.49, d, 6.3	4.51, d, 6.3	
2^{\prime}	5.73, t, 6.3	5.39, br t, 6.3	5.50, br t, 6.6	5.40, br t, 6.6	5.48, br t, 6.3	5.37, br t, 6.3	5.37, br t, 6.3	
$\frac{4}{3}$	1.991 m	2.01, m	2.13, m	2.03, m	2.11, m	2.03, m	2.03, m	
	1.971 m	1.95, m	2.10, m	2.02, m	2.09, m	1.97 m	2.01, m	
6^{\prime}	4.98, t, 6.6	4.98, br t, 5.7	5.11, br t, 5.7	5.00, br t, 6.3	5.09, br t, 6.3	4.99, br t, 6.9	4.99, br t, 6.6	
$\frac{8^{\prime}}{9^{\prime}}$	1.57 , s	1.57 , s	1.69, s	1.57, s	1.66 , s	1.57 , s	1.57, s	
	1.60, s	1.64 , s	1.78, s	1.68, s	1.73, s	1.66 , s	1.65, s	
10'	1.50, s	1.50, s	1.62, s	1.51, s	1.59, s	1.51, s	1.51 , s	
$1-OAc$	2.38, s							
$1-OH$		12.85, s	12.72, s	12.71 , s		12.55 , s		
$3-OH$		7.86, br s						
$7-OH$			7.03 , br s					
$1-OAC$					2.47, s		2.40, s	
$3-OAC$				2.23, s	2.27, s			
7-OAc						2.22, s	2.19, s	

([Fig. 3](#page-6-0)), while compound 15 was acetylated in the usual manner with Ac₂O in pyridine to give a crystalline monoacetate **18** ([Fig. 2\)](#page-5-0). Both 20 and 18 were subjected to X-ray diffraction analysis whose results supported the structures of 14 and 15.

Cochinchinone L (13) was isolated as a yellow powder, with a molecular ion peak at m/z 422.1718 [M]⁺ in the HRMS, corresponding to a molecular formula of $\mathsf{C}_{25}\mathsf{H}_{26}\mathsf{O}_{6}$. The $^{1}\mathsf{H}$ and $^{13}\mathsf{C}$ NMR spectral data of 13 ([Tables 3 and 4\)](#page-3-0) showed characteristic signals similar to those of 7-geranyloxy-1,3-dihydroxyxanthone (14) ,¹⁴ except that the hydroxyl group at C-1 of 14 was replaced by an acetoxy group in 13, which appeared as a proton singlet signal at δ 2.38. The higher field chemical shift of C-1 (δ 151.2) and lower field resonances of protons and carbons of C-2 (δ_H 6.43 (d, J 2.1 Hz, H-2); δ_C 108.3) and C-4 (δ_H 6.57 (d, J 2.1 Hz, H-4); δ_C 101.3) as compared to those of 14 (δ_H 6.22 (d, J 1.8 Hz, H-2); δ_C 98.5 and δ_H 6.23 (d, J 1.8 Hz, H-4); δ_C 94.4) supported the title structure [\(Tables](#page-3-0) [3 and 4](#page-3-0)). The 4 J HMBC correlations of methyl protons of the acetoxy group (δ 2.38) with C-1 (δ 151.2) supported its location at C-1, which was also confirmed by NOESY cross-peak between acetoxy moiety with H-2. Cochinchinone L (13), an acetylated analogue of

^a May be interchangeable.

14, was therefore assigned as 1-acetoxy-3-hydroxy-7-geranyloxyxanthone.

The isolated compounds were evaluated for their antibacterial activities against both Gram-positive bacteria: Bacillus subtilis, Staphylococcus aureus, Enterococcus faecalis TISTR 459, Methicillin-Resistant S. aureus (MRSA) ATCC 43300, Vancomycin-Resistant E. faecalis (VRE) ATCC 51299 and Gram-negative bacteria: Salmonella typhi, Shigella sonei, and P. aeruginosa. All compounds were also subjected to antifungal assay against Candida albicans. The results showed that most of the isolated compounds (4–6 and 13– 15) exhibited strong antibacterial activity specifically against P. aeruginosa ([Table 5\)](#page-6-0). Interestingly, only compounds 9 and 10 exhibited strong activity against C. albicans. It is important to note that compound 9 also exhibited broad spectrum antimicrobial activity against all Gram-positive bacteria. Nearly all compounds, which are active against P. aeruginosa are the 1,3,7-trihydroxy xanthones (4 and 5) or 1,3,7-trioxygenated xanthones with an oxygeranyl side chain either at C-3 or C-7 and dihydroxyl groups (14–15) whose indicated structures might contribute to the strong antibacterial activity specifically against P. aeruginosa. However, when the free hydroxyl group of the 1,3,7-trihydroxy xanthone was cyclized onto the isoprenyl or geranyl side chain to form a chromane or chromene ring, the antibacterial activity against P. aeruginosa decreased drastically as shown in compounds 1, 2, and 3a. In other previous reports, it was suggested that hydroxy xanthones play important roles in biological activity such as antibacterial, 3.5α glucosidase inhibitory,²⁰ anti-tumor,^{21,22} and anti-inflammatory effects.²³ Because compounds **14** and **15** are the major components obtained from this plant, this prompts us to modify their structures for the structure–activity relationships (SARs). To investigate whether the free hydroxyl group was responsible for antibacterial activity, the acetylation with acetic anhydride in pyridine was therefore applied to 14 and 15. Four acetylated geranyloxyxanthone derivatives: 3-acetoxy-7-geranyloxy-1-hydroxyxanthone (16) and 1,3-diacetoxy-7-geranyloxyxanthone (17) were obtained from 14, whereas 7-acetoxy-3-geranyloxy-1-hydroxy-xanthone (18) and 1,7-diacetoxy-3-geranyloxy-xanthone (19) were obtained from 15. The assignments of the $^1\mathrm{H}$ and $^{13}\mathrm{C}$ NMR spectral data of **16–19** are summarized in [Tables 3 and 4](#page-3-0). The antibacterial activity of acetylated geranyloxyxanthone derivatives 16–19 and dibrosylate 20 was evaluated. All of them showed strong anti-P. aeruginosa [\(Table](#page-6-0) [5\)](#page-6-0). This implied that the free hydroxyl groups should not be responsible for the inhibition of P. aeruginosa. For further investigation of the role of an oxygeranyl side chain, a 1,3,7-trihydroxyxanthone (21) obtained from the dried fruits of C. $cochinchine²⁴$ $cochinchine²⁴$ $cochinchine²⁴$ was tested against P. aeruginosa. It was inactive against Gram-negative bacteria as shown in [Table 5](#page-6-0). From these results, it can be suggested that the geranyl side chain is necessary for anti-P. aeruginosa activity. Moreover, mixtures of compound 4 with compound 5, and compound 14 with compound 15 were subjected to antimicrobial assay. Interestingly, the mixture of compounds 14 and 15 significantly increased the antibacterial activity against MRSA compared with the pure forms as indicated by the lower MIC values shown in [Table 5.](#page-6-0) The mixture of compounds 4 and 5, on the other hand, did not show any significant differences for antibacterial activity compared to the pure forms as also shown in [Table 5.](#page-6-0) Therefore, it may be proposed that the 1,3,7-trihydroxyxanthone with the isoprenyl or geranyl side chain at C-2 and C-4 in

Figure 2. ORTEP plot of a monoacetate 18

Figure 3. ORTEP plot of a dibrosylate 20.

4 and 5, respectively, and 1,3,7-trioxygenated xanthone with the geranyl side chain at C-3 or C-7 in 13–19 are essential for their antibacterial activity against P. aeruginosa. Therefore, 1,3,7-trihydroxyxanthones (4 and 5) and 1,3,7-trioxygenated xanthone with geranyl side chain (13–19) should be considered as potent candidates as anti-P. aeruginosa.

We further studied the possible mode of action of compounds 4 and 13 against P. aeruginosa by observing the bacteria cell morphology through scanning electron microscopy (SEM) at 3, 6, 9, and 15 h after applying compounds 4 and 13. From the SEM results (see [Figs 4](#page-7-0) [and 5\)](#page-7-0), it was clearly indicated that the cell morphology of P. aeruginosa, when treated with compounds 4 and 13, started to deform at 3 h onward, and at 15 h, most cells were completely deformed whose

results was correlated to their strong antibacterial activity (Table 5). Therefore, it can be suggested that compounds 4 and 13 may interact with or damage the cell wall of P. aeruginosa as seen by the formation of pores on the cell wall of P. aeruginosa ([Figs 4 and 5\)](#page-7-0).

3. Experimental

3.1. General experimental procedures

Melting points were determined on a Fisher–John melting point apparatus. Optical rotations were measured on a JASCO P-1020 digital polarimeter. UV and IR spectra were recorded on SPECORD S 100 (Analytikjena) and Perkin–Elmer FTS FT-IR spectrophotometer,

Table 5 Antimicrobial activity (MIC, μ g/ml) of 1-21

No.		Antibacterial activity								
		Gram-positive bacteria ^a					Gram-negative bacteria ^b			
	B. subtilis	S. aureus	E. faecalis	MRSA	VRE	S. typhi	S. sonei	P. aeruginosa	$C.$ albicans C	
$\mathbf{1}$	>300	>300	>300	>300	>300	>300	>300	>300	>300	
2	300	300	>300	300	300	>300	>300	300	75	
3a	75	>300	300	>300	>300	>300	>300	>300	300	
4	150	150	150	9.37	150	>150	>150	4.7	75	
5	150	75	150	37.5	75	>150	>150	4.7	37.5	
$4 + 5$	75	150	75	9.37	150	>150	>150	4.7	75	
6	$>\!\!150$	>150	>150	150	150	$>\!\!150$	>150	4.7	150	
7	150	300	300	150	150	>300	>300	150	300	
8	300	75	150	75	150	>300	>300	300	150	
9	9.37	9.37	9.37	9.37	9.37	>300	>300	18.75	2.4	
10	18.75	37.5	37.5	37.5	37.5	>300	>300	37.5	4.7	
11	>300	300	>300	150	150	>300	>300	>300	300	
12	75	>300	300	150	150	>300	>300	>300	>300	
13	150	>150	150	37.5	150	>150	>150	4.7	18.75	
14	150	>150	150	18.7	150	>150	>150	4.7	75	
15	150	150	150	9.3	150	>150	>150	4.7	37.5	
$14 + 15$	75	37.5	37.5	4.7	37.5	>150	>150	4.7	37.5	
16	150	>150	150	18.75	150	>150	>150	4.7	150	
17	75	>150	75	37.5	150	>150	>150	4.7	18.75	
18	>150	>150	150	18.75	150	>150	>150	4.7	150	
19	>150	>150	>150	37.5	150	>150	>150	4.7	150	
20	>150	>150	150	300	>300	>150	>150	4.7	>300	
21	37.5	75	>300	150	300	>300	>300	300	300	

a B. subtilis, S. aureus, E. faecalis TISTR 459, Methicillin-Resistant S. aureus (MRSA) ATCC 43300, Vancomycin-Resistant E. faecalis (VRE) ATCC 51299.

^b S. typhi, S. sonei, and P. aeruginosa.

^c C. albicans.

Figure 4. Scanning electron microscopy of cell morphology of P. aeruginosa treated with compound 4 at different time.

respectively. The 1 H and 13 C NMR spectra were recorded on 300 MHz Bruker FTNMR Ultra Shield^{TM} spectrometer in CDCl₃ with TMS as the internal standard. Chemical shifts are reported in δ (ppm) and coupling constants (*J*) are expressed in hertz. EI and HREI mass spectra were measured on a Kratos MS 25 RFA spectrometer. Quick column chromatography (QCC) and column chromatography (CC) were carried out on silica gel 60 $F₂₅₄$ (Merck) and silica gel 100 (Merck), respectively. All the bacteria images were viewed with a JSM-5800LV, JEOL SEM (scanning electron microscope).

3.2. Plant material

The resin of C. cochinchinense was collected in October 2003 at Prince of Songkla University, Hat-Yai campus, whereas the green fruits of C. cochinchinense were collected in October 2007 at Kaun Kha Long district, Satun Province, Southern part of Thailand. Botanical identification was achieved through comparison with a voucher specimen No. SL-1 (PSU) in the herbarium of Department of Biology, Prince of Songkla University, Songkhla, Thailand.

3.3. Isolation and extraction

The resin of C. cochinchinense (87.75 g) was extracted with CH_2Cl_2 (2×2.0 L, for a week) at room temperature and was evaporated under reduced pressure to afford a deep green crude $CH₂Cl₂$ extract (47.04 g), which was subjected to QCC (Quick column chromatography) on silica gel (Merck 60 F_{254}) using hexane as a first eluent and then increasing the polarity with acetone to give 16 fractions (FR1–FR16). Fractions FR4 and FR5 were separated by CC eluting with a gradient of acetone–hexane to give eight subfractions (FR4A–FR4H) and 8 (150.2 mg) (R_f (15% acetone–hexane (three runs)) 0.39). Subfraction FR4B was further purified by CC and eluted with a gradient of acetone–hexane to give 7 (31.4 mg) $(R_f (15\% \text{ acetone} - \text{hexane}) 0.31)$ and 12 (56.5 mg) $(R_f (15\% \text{ acetone} - \text{hexane}))$ hexane) 0.28). Fractions FR6 and FR7 were separated by QCC and eluted with a gradient of CH_2Cl_2 -hexane to give six subfractions (FR6A–FR6F). Subfraction FR6B was further separated by QCC eluting with a gradient of acetone–hexane to give 6 (35.7 mg) $(R_f (15\%)$ acetone–hexane (three runs)) 0.41), **8** (83.2 mg) $(R_f (15\%)$ acetone–hexane (three runs)) 0.39), and 11 (1.8 mg) $(R_f (15\%)$ acetone–hexane (three runs)) 0.24). Subfraction FR6E was purified by CC on reversed-phase silica gel C-18 eluting with MeOH to give 4 (849.4 mg) $(R_f$ (MeOH) 0.54) and 5 (1.25 g) $(R_f$ (MeOH) 0.62). Fractions FR8 and FR9 were separated by QCC and eluted with a gradient of CH₂Cl₂-hexane to afford **5** (551.3 mg) (R_f (60% CH₂Cl₂hexane (three runs)) 0.05), **7** (18.2 mg) (R_f (60% CH₂Cl₂–hexane (three runs)) 0.29), **8** (116.6 mg) (R_f (60% CH₂Cl₂–hexane (three runs)) 0.61), and 9 (148.7 mg) (R_f (60% CH₂Cl₂–hexane (three runs)) 0.10). Fractions FR10 and FR11 were separated by QCC and eluted with a gradient of acetone–hexane to give seven subfractions (FR10A–FR10G) and **7** (25.9 mg) $(R_f (15\%)$ acetone–hexane) 0.31). Subfraction FR10B was further purified by CC on silica gel C-18 and eluted with MeOH to furnish 1 (8.5 mg) (R_f (MeOH) 0.36). The mixture of 1 and 1a could not be separated due to the very low amount of 1a. The purity of 1 is much more than 90% base on a single spot on TLC and the integral area signal in the ¹H NMR spectrum of 1. Subfraction FR10D was separated by QCC eluting with a gradient of acetone–hexane to give six subfractions (FR10D1–FR10D6). Subfraction FR10D2 was further separated by CC and eluted with a gradient of acetone–hexane to give five subfractions (FR10D2A–FR10D2E), $2(1.8 \text{ mg}) (R_f(20\% \text{ acetone} - \text{hexane}))$

Figure 5. Scanning electron microscopy of cell morphology of P. aeruginosa treated with compound 13 at different time.

0.36), 4 (23.1 mg) (R_f (20% acetone–hexane) 0.22), 5 (34.2 mg) (R_f (20% acetone–hexane) 0.17), and an inseparable mixture of 2 and 3 (20.2 mg) (R_f (20% acetone–hexane) 0.35). The mixture was separated by acetylation with Ac_2O (0.1 ml) in pyridine (2.0 ml) and stirred overnight at room temperature to give a yellow gum, which was further purified by CC eluting with 70% CHCl₃-hexane to give acetylated derivatives $2a(5.1 \text{ mg})(R_f(70\% \text{CHCl}_3-\text{hexane})0.68)$ and **3a** (18.9 mg) $(R_f(70\% \text{CHCl}_3\text{-} \text{hexane}) 0.40)$, respectively. Subfraction FR10D2E was purified by CC and eluted with 80% CH₂Cl₂-hexane to give 8 (16.7 mg) (R_f (80% CH₂Cl₂–hexane) 0.91), **10** (5.0 mg) (R_f (80%) CH₂Cl₂–hexane) 0.85), and **11** (1.5 mg) (R_f (80% CH₂Cl₂–hexane) 0.38).

Air-dried green fruits of C. cochinchinense (5.5 kg) were extracted with CH_2Cl_2 (2×20 L, for a week) at room temperature and was evaporated under reduced pressure to afford a deep green crude CH_2Cl_2 extract (40.04 g), which was subjected to QCC on silica gel using hexane as a first eluent and increasing polarity with EtOAc to give nine fractions (FS1–FS9). Fraction FS5 was purified by CC eluting with pure CHCl₃ to give **14** (1.88 g) (R_f (70%) CHCl3–hexane) (three runs) 0.46). Fraction FS6 was further separated by CC eluting with pure CHCl₃ to furnish six subfractions (FS6A–FS6F), **14** (2.10 g) (R_f (70% CHCl₃–hexane) (three runs) 0.46) and **15** (490.2 mg) (R_f (70% CHCl₃–hexane) (three runs) 0.39). Subfraction FS6B was further purified by CC eluting with a gradient of acetone–hexane to give 13 (53.3 mg) (R_f (15% acetone– hexane (three runs)) 0.17).

3.3.1. Cochinchinone I (1)

Yellow needle single crystals, mp 160–162 °C; UV (CHCl₃) λ_{max} $(\log \varepsilon)$ 261 (4.01), 297 (4.31), 342 (3.69), 393 (3.46) nm; IR (neat) $\nu_{\rm max}$ 3406, 1650, 1612 cm $^{-1}$; HRMS *m*/z 446.2279 for C₂₈H₃₀O₅ (calcd 446.2093). EIMS m/z (rel int.): 446 [M]⁺ (76), 431 (100), 377

(73), 363 (76), 323 (26), 307 (15), 295 (13), 137 (5), 69 (6). For ¹H and ¹³C NMR spectroscopic data, see [Table 1.](#page-1-0)

3.3.2. Cochinchinone $J(2)$

Yellow viscous oil; $[\alpha]_D^{25}$ –69.8 (c 0.08, CHCl₃); UV (CHCl₃) λ_{max} $(\log \varepsilon)$ 243 (3.90), 289 (4.10), 298 (4.13), 320 (3.68), 351 (3.47), 391 (3.33) nm; IR (neat) v_{max} 3397, 1649, 1613 cm⁻¹; HRMS m/z 446.2092 for C₂₈H₃₀O₅ (calcd 446.2093). EIMS m/z (rel int.): 446 $[M]^{+}$ (11), 363 (100), 307 (18), 69 (5). For ¹H and ¹³C NMR spectroscopic data, see [Table 1.](#page-1-0)

3.3.3. Acetylated derivative of cochinchinone $K(3a)$

Yellow powder, mp 85–87 °C; UV (CHCl₃) λ_{max} (log ε) 243, 271, 303, 326, 383 nm; IR (neat) v_{max} 3392, 1709, 1648, 1612 cm⁻¹; HRMS m/z 490.2355 for C₃₀H₃₄O₆ (calcd 490.2355). EIMS m/z (rel int.): $490 \, [\text{M}]^+$ (53), 473 (53), 419 (44), 405 (31), 365 (100), 323 (52), 311 (41), 267 (51), 69 (24). For ¹H and ¹³C NMR spectroscopic data, see [Table 1.](#page-1-0)

3.3.4. Cochinchinone L (13)

Yellow powder, mp 114-116 °C; UV (CHCl₃) λ_{max} (log ε) 248 (4.59), 273 (4.03), 305 (4.13), 354 (3.80) nm; IR (neat) v_{max} 3237, 1774, 1728, 1628 cm⁻¹; HRMS m/z 422.1718 for C₂₅H₂₆O₆ (calcd 422.1729). EIMS m/z (rel int.): 422 [M]⁺ (1), 286 (40), 244 (100), 187 (4), 81 (9), 69 (28). For ¹H and ¹³C NMR spectroscopic data, see [Tables 3 and 4](#page-3-0).

3.4. Acetylation of 14 and 15

Compound **14** (82.5 mg) was treated with Ac_2O (2.5 ml) in pyridine (2.0 ml) and stirred for 6 h at room temperature. The reaction mixture was diluted with water and extracted with $CH₂Cl₂$. The combined organic extract was washed with 10% HCl and then washed with water again. After the organic solvent was removed, the resulting residue was dried over anhydrous $Na₂SO₄$. Chromatography over silica gel yielded a pale yellow powder of 16 (80.6 mg) (R_f (5% acetone–hexane (three runs)) 0.44).

Compound 14 (200.5 mg) was treated with Ac₂O (6.0 ml) in pyridine (3.0 ml) and stirred overnight at room temperature. Chromatography over silica gel yielded a pale yellow powder of 16 (10.6 mg) $(R_f (5\%)$ acetone–hexane (three runs)) 0.44) and 17 (177.8 mg) $(R_f (15\% \text{ acetone} - \text{hexane} (\text{three runs})) 0.46)$.

Compound 15 (85.5 mg) was treated with Ac_2O (2.5 ml) in pyridine (2.0 ml) and stirred for 6 h at room temperature. Chromatography over silica gel yielded a pale yellow powder of 18 (83.7 mg) $(R_f (5\%)$ acetone–hexane (three runs)) 0.32), which was further recrystallized from MeOH–acetone (1:99 v/v) to give yellow single crystals.

Compound 15 (190.0 mg) was treated with Ac_2O (6.0 ml) in pyridine (3.0 ml) and stirred overnight at room temperature. Chromatography over silica gel yielded a pale yellow powder of 18 (8.7 mg) $(R_f (5\%)$ acetone–hexane (three runs)) 0.32) and 19 (170.0 mg) $(R_f (15\%)$ acetone–hexane (three runs)) 0.37).

3.4.1. 3-Acetoxy-7-geranyloxy-1-hydroxyxanthone (16)

Yellow powder, mp 94–95 °C; UV (CHCl₃) λ_{max} (log ε) 239 (4.40), 262 (4.66), 289 (3.97), 379 (3.93) nm; IR (neat) v_{max} 3429, 1768, 1649, 1612 cm⁻¹; HRMS m/z 422.1725 for C₂₅H₂₆O₆ (calcd 422.1729). EIMS m/z (rel int.): 422 [M]⁺ (1), 286 (43), 244 (100), 187 (4), 81 (9), 69 (24). For ¹H and ¹³C NMR spectroscopic data, see [Tables 3 and 4](#page-3-0).

3.4.2. 1,3-Diacetoxy-7-geranyloxyxanthone (17)

Yellow powder, mp 96–97 °C; UV (CHCl₃) λ_{max} (log ε) 253 (4.59), 300 (3.45), 361 (3.86) nm; IR (neat) v_{max} 3429, 1776, 1656, 1624 cm⁻¹; HRMS m/z 464.1838 for C₂₇H₂₈O₇ (calcd 464.1835). EIMS m/z (rel int.): 464 [M]⁺ (2), 328 (3), 286 (58), 244 (100), 187 (5), 81 (17), 69 (36). For ¹H and ¹³C NMR spectroscopic data, see [Tables 3 and 4](#page-3-0).

3.4.3. 7-Acetoxy-3-geranyloxy-1-hydroxy-xanthone (18)

Yellow powder, mp 104-106 °C; UV (CHCl₃) λ_{max} (log ε) 243 (4.44), 257 (4.52), 310 (4.29), 358 (3.82) nm; IR (neat) ν_{max} 3429, 1758, 1665, 1607 cm⁻¹; HRMS m/z 422.1726 for C₂₅H₂₆O₆ (calcd 422.1729). EIMS m/z (rel int.): 422 [M]⁺ (5), 286 (16), 244 (100), 187 (2), 81 (16), 69 (56). For ¹H and ¹³C NMR spectroscopic data, see [Tables 3 and 4](#page-3-0).

3.4.4. 1,7-Diacetoxy-3-geranyloxyxanthone (19)

Yellow powder, mp 85–87 °C; UV (CHCl₃) λ_{max} (log ε) 246 (4.61), 275 (4.01), 302 (4.28), 334 (3.86) nm; IR (neat) v_{max} 3453, 1770, 1655, 1629 cm⁻¹; HRMS m/z 464.1834 for C₂₇H₂₈O₇ (calcd 464.1835). EIMS m/z (rel int.): 464 [M]⁺ (4), 328 (4), 286 (32), 244 (100), 187 (2), 81 (24), 69(73). For ¹H and ¹³C NMR spectroscopic data, see [Tables 3 and 4.](#page-3-0)

3.5. Brosylation of 14

Compound 14 (40.0 mg, 105.14 mmol) was stirred overnight at room temperature with p-bromobenzenesulfonyl chloride $(40.30 \text{ mg}, 190.2 \text{ mmol})$ and $K_2CO_3(44.1 \text{ mg}, 315.4 \text{ mmol})$ in CH₂Cl₂ (3 ml). After the reaction was complete, water (10 ml) was added to the reaction mixture. The resulting solution was then extracted with $CH₂Cl₂$ (10 ml, three times). The combined organic extract was dried over anhydrous sodium sulfate and evaporated under reduced pressure to give a crude extract, which was further purified by column chromatography over silica gel eluting with 5% acetone–hexane to yield the dibrosylate **20** (75.2 mg) $(R_f (5\%)$ acetone–hexane (three runs)) 0.20). Yellow needle-shaped single crystals of 20 were obtained after recrystallization from CH₃OH–CHCl₃ (1:4 v/v), mp 106–108 °C. δ_H (300 MHz, CDCl₃+CD₃OD) 7.82 (dd, J 9.0, 2.1 Hz, H-2", H-6"), 7.68 (d, J 2.4 Hz, H-4), 7.65 (dd, J 8.7, 2.1 Hz, H-3", H-5", H-2"", H-6""), 7.63 (dd, J 8.7, 2.1 Hz, H-3"", H-5""), 7.50 (d, J 2.4 Hz, H-8), 7.32 (d, J 2.4 Hz, H-2), 7.26 (d, J 9.0 Hz, H-5), 7.19 (dd, J 9.3, 2.4 Hz, H-6), 5.44 (br t, J 6.6, H-2'), 5.04 (br t, J 6.3 Hz, H-6'), 4.56 (d, J 6.6 Hz, H-1'), 2.07 (m, 2H-5'), 2.05 (m, 2H-4'), 1.71 (s, 3H-9'),1.60 (s, 3H-8'), 1.54 (s, 3H-10'); δ _C(75 MHz, CDCl₃+CD₃OD) 181.6 (C-9), 157.4 $(C-1)$, 155.9 $(C-7)$, 151.9 $(C-3)$, 149.7 $(C-4b)$, 148.6 $(C-4a)$, 142.1 $(C-3')$, 134.3 (C-1"), 133.4 (C-1"'), 133.1 (C-3", C-5"), 132.5 (C-3"", C-5"'), 131.8 (C-7'), 130.6 (C-4"), 130.2 (C-2", C-6"), 130.1 (C-4""), 129.8 (C-2", C-6"'), 124.9 (C-6), 123.7 (C-6'), 122.4 (C-8a), 118.8 (C-5), 118.6 (C-2'), 114.2 (C-9a), 112.8 (C-4), 111.1 (C-2), 106.8 (C-8), 65.6 (C-1'), 39.5 (C-4'), 26.2 (C-5'), 25.5 (C-8'), 17.5 (C-10'), 16.6 (C-9'). EIMS m/z (rel int.): 816 [M-2]⁺ (1), 683 (7), 681 (14), 679 (7), 619 (5), 617 (10), 615 (5), 461 (55), 463 (56), 399 (41), 397 (41), 371 (19), 369 (19), 357 (34), 355 (34), 244 (27), 229 (84), 215 (100),186 (16),157 (53),155 (53),131 (11), 108 (13), 76 (16).

3.6. X-ray crystallographic studies of cocrystal of 1 and 1a, monoacetate 18, and dibrosylate 20

Crystallographic data were collected at 100.0 (1) K with the Oxford Cryosystem Cobra low-temperature attachment. The data were collected using a Bruker Apex2 CCD diffractometer with a graphite monochromated Mo Ka radiation at a detector distance of 5 cm using APEX2.^{[25](#page-10-0)} The collected data were reduced using SAINT program,^{[25](#page-10-0)} and the empirical absorption corrections were performed using SADABS program[.25](#page-10-0) The structures were solved by direct methods and refined by least-squares using the SHELXTL software package^{[26](#page-10-0)} All non-hydrogen atoms were refined anisotropically, whereas all H atoms were placed in calculated positions with an O–H distance of 0.82 Å and C–H distances in the range 0.93–0.98 Å after checking their positions in the difference map. The U_{iso} values were constrained to be 1.5 U_{eq} of the carrier atoms for methyl H atoms and $1.2U_{eq}$ for hydroxyl and the other H atoms. The final refinement converged well. Materials for publication were prepared using SHELXTL^{[26](#page-10-0)} and PLATON.^{[27](#page-10-0)}

Crystal data for cocrystal of 1 and 1a: $C_{28.60}H_{32.40}O_5$, M=465.15, 0.60 \times 0.19 \times 0.10 mm³, monoclinic, P2₁/c, a=21.6161(6) Å, b= 5.3826(2) Å, c=22.8296(8) Å, α =90.00°, β =121.267°(2), γ =90.00°, V=2270.44(13) \AA^3 , Z=4, D_x=1.334 Mg m⁻³, μ (Mo Kα)=2.512 mm⁻¹, 29,338 reflection measured, 6634 unique reflections, $R=0.0995$, R_{w} =0.2794.

Crystal data for **18**: $C_{25}H_{26}O_6$, M=422.46, 0.58 \times 0.27 \times 0.06 mm³, orthorhombic, $P2_12_12_1$, $a=7.3280(2)$ Å, $b=13.0634(5)$ Å, $c=45.6190$ (18) Å, $\alpha = \beta = \gamma = 90.00^{\circ}$, V=4367.0(3) Å³, Z=4, D_x=1.285 Mg m⁻³, μ (Mo K α)=0.091 mm $^{-1}$, 81,117 reflection measured, 4877 unique reflections, $R=0.1258$, $R_w=0.3299$.
Crystal data for **20**:

Crystal data for **20**: $C_{35}H_{30}Br_2O_9S_2$, *M*=818.53, $0.60\times0.18\times0.03$ mm³, triclinic, *P*-1, *a*=9.0779(4) Å, *b*=19.3468(7) Å, c=20.7939(7) Å, α =108.160(2)°, β =91.755(2)°, γ =90.046(2)°, $V = 3468.3(2) \text{ Å}^3$, $Z = 4$, , $Z=4$, $D_x=1.568 \text{ Mg m}^{-3}$, $\mu(\text{Mo K}\alpha)$ = 2.512 mm^{-1}, 41,231 reflection measured, 12,219 unique reflections, $R = 0.0817$, $R_w = 0.2121$.

The crystallographic-information files for cocrystal of 1 and 1a, 18, and 20 have been deposited in the Cambridge Crystallographic Data Centre as CCDC694467, CCDC689924, and CCDC689378, respectively. These data can be obtained free of charge via [http://](http://www.ccdc.cam.ac.uk/data_request/cif) www.ccdc.cam.ac.uk/data_request/cif, or by e-mailing [data_](mailto:data_request@ccdc.cam.ac.uk) [request@ccdc.cam.ac.uk,](mailto:data_request@ccdc.cam.ac.uk) or by contacting the Cambridge Crystallographic Data Centre, 12, Union Road, Cambridge CB2 1EZ, UK; fax: þ44 1223 336033.

3.7. Antimicrobial activity

3.7.1. Antibacterial assay

The isolated compounds from the resin and green fruits of C. cochinchinense were tested against both Gram-positive and Gramnegative bacteria: B. subtilis, S. aureus, TISTR517, E. faecalis TISTR459, Methicillin-Resistant S. aureus (MRSA) ATCC43300, Vancomycin-Resistant E. faecalis (VRE) ATCC 51299, Streptococcus faecalis, S. typhi, S. sonei and P. aeruginosa. The microrganisms were obtained from the culture collections, Department of Industrial Biotechnology and Department of Pharmacognosy and Botany, PSU, except for the TISTR and ATCC strains, which were obtained from Microbial Research Center (MIRCEN), Bangkok, Thailand. The antibacterial assay employed was the same as described in Boonsri et al. 5 Vancomycin, which was used as a standard, showed antibacterial activity against Vancomycin-Resistant E. faecalis (VRE) ATCC 51299 at 75.0 μ g/ml.

3.7.2. Antifungal assay

C. albicans was obtained from Department of Pharmacognosy and Botany, PSU. The antifungal assay employed was the same as described in Boonsri et al.⁵ by using amphotencin B as a positive control.

3.7.3. Antibacterial assay against P. aeruginosa and SEM analysis for treated cells

Antibacterial activity against P. aeruginosa and the potential mode of action of compounds 4 and 13 were investigated by the same antibacterial assay. The cells of P. aeruginosa were sampled at 0, 3, 6, and 15 h for analysis by SEM.

Acknowledgements

We are greatly indebted to the Thailand Research Fund (TRF) for research grant (RSA5280033). Financial support from the Center of Excellence for Innovation in Chemistry (PERCH-CIC), Commission on Higher Education, Ministry of Education is gratefully acknowledged. N.B. thanks the Development and Promotion of Science and Technology Talents Project for fellowship and Graduate School, Prince of Songkla University for financial supports. The authors also thank the Koshinocorporation Group, Japan and Universiti Sains Malaysia for the Research University Golden Goose grant No. 1001/ PFIZIK/811012. N.B. thanks Asst. Prof. Dr. Surat Laphookhieo for supplying the authentic sample of 1,3,7-trihydroxyxanthone. The authors also would like to thank the Scientific Equipment Center, Prince of Songkla University for the SEM analysis.

References and notes

- 1. Smitinand, T. Thai Plant Names; Prachachon: Bangkok, 2001; p 152.
2. Vo. V. V. A Dictionary of Medicinal Plants in Vietnam: Y Hoc: HoCh
- 2. Vo, V. V. A Dictionary of Medicinal Plants in Vietnam; Y Hoc: HoChiMinh City, 1997; Vol. 435.
- 3. Boonnak, N.; Karalai, C.; Chantrapromma, S.; Ponglimanont, C.; Hoong-Kun, F.; Kanjana-Opas, A.; Laphookhieo, S. Tetrahedron 2006, 62, 8850–8859.
- 4. Boonnak, N.; Karalai, C.; Chantrapromma, S.; Ponglimanont, C.; Kanjana-Opas, A.; Chantrapromma, K.; Hoong-Kun, F. Can. J. Chem. 2007, 85, 341–345.
- 5. Boonsri, S.; Karalai, C.; Ponglimanont, C.; Kanjana-Opas, A.; Chantrapromma, K. Phytochemistry 2006, 67, 723–727.
- 6. Mahabusarakam, W.; Nuangnaowarat, W.; Taylor, W. C. Phytochemistry 2006, 67, 470–474.
- 7. Deachathai, S.; Mahabusarakam, W.; Phongpaichit, S.; Taylor, W. C. Phytochemistry 2005, 66, 2368–2375.
- 8. Suksamrarn, S.; Komutiban, O.; Ratananukul, P.; Chimnoi, N.; Lartpornmatulee, N.; Suksamrarn, A. Chem. Pharm. Bull. 2006, 54, 301–305.
- 9. Suksamrarn, S.; Suwannapoch, N.; Phakhodee, W.; Thanuhiranlert, J.; Ratananukul, P.; Chimnoi, N.; Suksamrarn, A. Chem. Pharm. Bull. 2003, 51, 857–859.
- 10. Molinar-Toribio, E.; González, I.; Ortega-Barría, E.; Capson, T. L.; Coley, P. D.; Kursar, T. A.; McPhail, K.; Cubilla-Rios, L. Pharm. Biol. 2006, 44, 550–553.
- 11. Engelhard, D. Bone Marrow Transplant 1998, 21, S78–80.
- 12. Collin, B. A.; Leather, H. L.; Wingard, J. R.; Ramphal, R. Clin. Infect. Dis. 2001, 33, 947–953.
- 13. Mencacci a, A.; Cenci, E.; Repetto, A.; Mazzolla, R.; Bistoni, F.; Aversa, F.; Aloisi,
- T.; Vecchiarelli. J. Infect. 2006, 53, 259–264. 14. Nguyen, L. H. D.; Harrison, L. J. Phytochemistry 1998, 50, 471–476.
- 15. Dechathai, S.; Mahabusarakam, W.; Phongpaichit, S.; Taylor, W. C.; Zhang, Y. J. Phytochemistry 2006, 67, 464–469.
- 16. Mahabusarakam, W.; Wiriyachitra, P.; Taylor, W. C. J. Nat. Prod. 1987, 50, 474– 478.
- 17. Delle Monache, F.; Botta, B.; Nicoletti, M.; De Barros Coelho, J. S.; Lyra, F. D.; De Andrade Lyra, F. D. J. Chem. Soc., Perkin Trans. 1 1981, 484–488.
- 18. Sadaquat, A.; Goundar, R.; Sotheeswaran, S.; Beaulieu, C.; Spino, C. Phyto-chemistry 2000, 53, 281–284.
- 19. Rattanaburee, S; Mahabusarakam, W. Chemical Constituents from the Twigs and Flowers of Cratoxylum cochinchinense, In 32nd Congress on Science and Technology of Thailand (STT32), Queen Sirikit National Convention Center, Bangkok, Thailand, Oct. 10–12, 2006.
- 20. Liu, Y.; Zou, L.; Ma, L.; Chen, W. H.; Wang, B.; Xu, Z. L. Bioorg. Med. Chem. 2006, 14, 5683–5690.
- 21. Pedro, M.; Cerqueira, F.; Sousa, M. E.; Nascimento, M. S. J.; Pinto, M. Bioorg. Med. Chem. 2002, 10, 3725–3730.
- 22. Yoshimi, N.; Matsunaga, K.; Katayama, M.; Yamada, Y.; Kuno, T.; Qiao, Z.; Hara, A.; Yamahara, J.; Mori, H. Cancer Lett. 2001, 163, 163–170.
- 23. Lin, C.-N.; Chung, M.-I.; Liou, S.-J.; Lee, T.-H.; Wang, J.-P. J. Pharm. Pharmacol. 1996, 48, 532–538.
- 24. Laphookhieo, S.; Maneerat, W.; Buatip, T.; Syers, J. K. Can. J. Chem. 2008, 86, 757–760.
- 25. Bruker. APEX2, SAINT and SADABS; Bruker AXS Inc.: Madison, WI, 2005.
- 26. Sheldrick, G. M. SHELXTL Version 5.1; Bruker AXS; USA: Madison, WI, 1998.
- 27. Spek, A. L. J. Appl. Crystallogr. 2003, 36, 7–13.